

# Cambridge IGCSE™

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**CO-ORDINATED SCIENCES****0654/52**

Paper 5 Practical Test

**May/June 2025****MARK SCHEME**

Maximum Mark: 60

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**Published**

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the May/June 2025 series for most Cambridge IGCSE, Cambridge International A and AS Level components, and some Cambridge O Level components.

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This document consists of **11** printed pages.

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptions for a question. Each question paper and mark scheme will also comply with these marking principles.

**GENERIC MARKING PRINCIPLE 1:**

Marks must be awarded in line with:

- the specific content of the mark scheme or the generic level descriptors for the question
- the specific skills defined in the mark scheme or in the generic level descriptors for the question
- the standard of response required by a candidate as exemplified by the standardisation scripts.

**GENERIC MARKING PRINCIPLE 2:**

Marks awarded are always **whole marks** (not half marks, or other fractions).

**GENERIC MARKING PRINCIPLE 3:**

Marks must be awarded **positively**:

- marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate
- marks are awarded when candidates clearly demonstrate what they know and can do
- marks are not deducted for errors
- marks are not deducted for omissions
- answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.

**GENERIC MARKING PRINCIPLE 4:**

Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

**GENERIC MARKING PRINCIPLE 5:**

Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).

**GENERIC MARKING PRINCIPLE 6:**

Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

**Science-Specific Marking Principles**

- 1 Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
- 2 The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
- 3 Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
- 4 The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.

**5 'List rule' guidance**

For questions that require ***n*** responses (e.g. State **two** reasons ...):

- The response should be read as continuous prose, even when numbered answer spaces are provided.
- Any response marked *ignore* in the mark scheme should not count towards ***n***.
- Incorrect responses should not be awarded credit but will still count towards ***n***.
- Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response.
- Non-contradictory responses after the first ***n*** responses may be ignored even if they include incorrect science.

**6 Calculation specific guidance**

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form (e.g.  $a \times 10^n$ ) in which the convention of restricting the value of the coefficient (a) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

**7 Guidance for chemical equations**

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

**Annotations guidance for centres**

Examiners use a system of annotations as a shorthand for communicating their marking decisions to one another. Examiners are trained during the standardisation process on how and when to use annotations. The purpose of annotations is to inform the standardisation and monitoring processes and guide the supervising examiners when they are checking the work of examiners within their team. The meaning of annotations and how they are used is specific to each component and is understood by all examiners who mark the component.

We publish annotations in our mark schemes to help centres understand the annotations they may see on copies of scripts. Note that there may not be a direct correlation between the number of annotations on a script and the mark awarded. Similarly, the use of an annotation may not be an indication of the quality of the response.

The annotations listed below were available to examiners marking this component in this series.

**Annotations**

Annotation	Meaning
	correct point or mark awarded
	incorrect point or mark not awarded
	benefit of the doubt given
	response is too vague or there is insufficient detail in response
	error carried forward applied
	information missing or insufficient for credit
	unclear response
	incorrect or insufficient point ignored while marking the rest of the response
	incorrect point or mark not awarded
	contradiction in response, mark not awarded

Annotation	Meaning
<b>SEEN</b>	point has been noted, but no credit has been given or blank page seen
	key point attempted / working towards marking point / incomplete answer / response seen but not credited / blank page seen
sup	refers to the supervisors' responses

Question	Answer	Marks
1(a)(i)	quality – clear and continuous outline ; size – at least half of the box and all drawing in the box ; detail – petals and carpel and stamen ;	3
1(a)(ii)	stigma correctly labelled ;	1
1(b)(i)	line drawn and measurement correct in mm ;	1
1(b)(ii)	measurement correct ;	1
1(b)(iii)	correct calculation ; to 2SF ;	2
1(b)(iv)	take more <b>readings</b> (of width) from different petals and calculate an average ;	1
1(c)	carbon dioxide and water ; carbon and hydrogen ;	2

Question	Answer	Marks
2(a)	one column / row headings for reagents ; column / row headings for observations / (final) colours for solutions <b>A</b> and <b>B</b> ;	2
2(b)(i)	<b>A</b> – yellow / green / orange / red for Benedict's ; <b>B</b> – purple for biuret ; <b>A</b> biuret and <b>B</b> Benedict's blue ; <b>A</b> and <b>B</b> both brown with iodine ;	4

Question	Answer	Marks
2(b)(ii)	<b>A</b> contains <b>reducing</b> sugar ; <b>B</b> contains protein ; neither contain starch ;	3

Question	Answer	Marks
3(a)	number of spatula loads in 20 cm <sup>3</sup> ; full set of values ; values increasing ;	3
3(b)(i)	axes correct way round and labelled with quantity and unit ; sensible linear scales with points covering $\geq 1/2$ grid and all points can be plotted ; points plotted correctly $\pm 1/2$ small square ;	3
3(b)(ii)	best-fit straight line ;	1
3(b)(iii)	as volume (of water) increases number of spatulas (of K <sub>2</sub> SO <sub>4</sub> ) increases ;	1
3(b)(iv)	correct value from graph ; 2 lines drawn one from each axis to line $\pm 1/2$ small square ;	2
3(c)	value calculated correctly ;	1
3(d)	use a graduated/volumetric pipette / burette for volume of water ; use mass of potassium sulfate ;	2

Question	Answer	Marks
4(a)(i)	2 temperatures ; both temperatures to nearest 0.5° ; solution higher temperature ;	3
4(a)(ii)	$\Delta T$ and $\Delta E$ calculated ;	1
4(b)(i)	(pale) blue ppt ; deep blue solution ;	2
4(b)(ii)	(aqueous) ammonia ;	1

Question	Answer	Marks
5(a)	diameter recorded for 20.0 cm height ; 5 results ; results increase down the table ; all results in mm <b>AND</b> 10 to 200 ;	4
5(b)	relationship correct ;	1
5(c)(i)	protractor / set square ;	1

Question	Answer	Marks
5(c)(ii)	<p>source of error identified ;  suggestion is appropriate ;  e.g.  (difficult to measure) crater is not circular / irregular / difficult to tell edge ;  measure several 'diameters' on the (same) crater and average ;</p> <p>OR  crater crumbles / sand moves when ball removed / diameter changes as removes ball / hands may make crater larger ;  use damp sand / use compasses/divider to determine diameter ;</p> <p>OR  hand wobbles when dropping ;  use steel ball bearing and electromagnet / hang from a string and cut string ;</p> <p>OR  difficult to measure diameter of crater <b>as</b> ruler cannot be laid on the sand ;  use compasses/divider/smaller rule ;</p> <p>OR  some of sand spills from the tray ;  refill / weigh the sand ;</p>	2
5(d)(i)	23 ;	1
5(d)(ii)	32 circled ;  27 is <b>anomalous / outlier</b> ;	2
5(e)(i)	33.6 ;	1
5(e)(ii)	no <b>and</b> ratio $V/d$ or $d/V$ is not constant ;	1

Question	Answer	Marks
6	<p>one mark from each section and any two others (if one section is omitted then max 6 etc.)</p> <p><b>apparatus</b> thermometer and temperature measured in plan ; measuring cylinder and volume/amount measured in plan ;</p> <p><b>method</b> hot water in beaker and wrap newspaper and measure temperature and time for a minimum of 2 thicknesses of paper ;</p> <p><b>measurements</b> temperature change after specified time/times or time for a specified temperature change ; repeat each number of layers to identify/exclude anomalies ; (minimum of) 5 layers ;</p> <p><b>control variables</b> initial temperature of the hot water ; volume of water ; temperature of the surroundings / room temperature ; size / shape of beaker ;</p> <p><b>conclusion</b> <b>calculate rate</b> temp change over time ; plot <b>graph</b> of rate (of cooling) against (number of) layers ; describe possible shapes of graph e.g. straight line – linear relationship / straight line through origin – proportional etc. ; when newspaper layers increase does rate (of cooling) increase / decrease or stay the same ;</p>	7